Phenol Esters of 2,3,5-Triiodobenzoic Acid

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> The 2,3,5-triiodobenzoic esters of thirty-seven phenols have been prepared by reacting the various phenols with 2,3,5-triiodobenzovl chloride.

 $\mathbf{P}_{\mathrm{HENOL}}$ ESTERS of 3,4,5-triiodobenzoic acid were reported (2) and the phenol esters of 2,3,5-triiodobenzoic acid have now been prepared.

EXPERIMENTAL

The phenols were used as obtained from commercial sources. The acid chloride was prepared by the method of Klemme and Hunter (1) except that cyclohexane was used for crystallization.

The method described in (3) was used except that 2 ml. of pyridine was added to the reaction mixture. Unless otherwise indicated in Table I, 1-butanol was used as the solvent.

LITERATURE CITED

- (1)Klemm, C.J., Hunter, J.H., J. Org. Chem. 5, 508 (1940). O'Donnell, D.C., Isaacs, V.A., Kiely, L.E., Millard, R.J., Welchlin, J.A., J. CHEM. ENG. DATA 8, 608 (1963). (2)
- O'Donnell, D.C., Kelley, J.K., Jr., O'Malley, R.F., Upham, (3)R.H., J. Am. Chem. Soc. 70, 1657 (1948).

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Phenol Used	M. ₽., ° C.	Yield, %	Formula	Iodine, %	
				Calcd.	Found
2-Aceto-1-naphthol	181.5 - 182.0	27	$C_{19}H_{11}O_{3}I_{3}$	57.0	57.02
Acetyl-p-methylaminophenol	172.4-173.8°	19	$C_{16}H_{12}O_{3}NI_{3}$	58.85	59.30
p-Benzylphenol	147.4 - 148.2	50	$C_{20}H_{13}O_{2}I_{3}$	57.16	57.44
p-Bromphenol	$162.8 - 163.8^{\circ}$	36	$C_{13}H_6O_2BrI_3$	58.14	57.73
Catechol	174.6-175.2°	42	$C_{20}H_{8}O_{4}I_{6}$	70.92	71.01
2-Chloro-5-hydroxytoluene	156.6 - 157.4	58	$C_{14}H_8O_2ClI_3$	60.98	61.43
o-Chlorophenol	$139.3 - 140.6^{\flat}$	64	$C_{13}H_6O_2ClI_3$	62.39	62.10
<i>m</i> -Chlorophenol	$130.6 - 132.3^{\circ}$	21	$C_{13}H_6O_2ClI_3$	62.39	62.17
p-Chlorothymol	$124.8 - 125.4^{\circ}$	52	$C_{17}H_{14}O_2ClI_3$	57.14	57.52
2,4-Dichloro-1-naphthol	$234.0 - 235.2^{d}$	37	$C_{17}H_7O_2Cl_2I_3$	54.81	54.48
2,6-Dichloro-4-nitrophenol	157.0 - 158.6	30	$C_{13}H_4O_4NCl_2I_3$	55.19	54.96
2,4-Dichlorophenol	$170.8 - 172.2^{b}$	32	$C_{13}H_5O_2Cl_2I_3$	59.05	59.12
2,4-Dihydroxybenzaldehyde	175.5 - 176.2	27	$C_{21}H_8O_5I_6$	69.10	68.58
2,2'-Dihydroxybinaphthyl-1,1'	259.4-260.4°	42	$C_{34}H_{16}O_4I_6$	60.92	61.25
2,7-Dihydroxynaphthalene	$218.0 - 220.8^{e}$	31	$C_{24}H_{10}O_{4}I_{6}$	67.76	67.87
2,5-Dihydroxytoluene	$268.0 - 269.8^{t}$	25	$C_{21}H_{10}O_4I_6$	70.00	69.90
3,5-Dinitro-o-cresol	171.0 - 171.8	18	$C_{14}H_7O_6N_2I_3$	56.00	56.49
2,4-Dinitrophenol	$154.6 - 155.8^{b}$	31	$C_{13}H_5O_6N_2I_3$	57.18	57.40
Eugenol	$123.5 - 125.0^{\circ}$	45	$C_{17}H_{13}O_{3}I_{3}$	58.94	58.81
Hydroquinone monobenzylether	137.7 - 138.6	45	$C_{20}H_{13}O_{3}I_{3}$	55.85	56.01
Hydroquinone monomethylether	$152.0 - 154.0^{b}$	36	$C_{14}H_9O_3I_3$	62.84	63.01
4-Hydroxy-1,2-dimethylbenzene	137.0 - 137.6	49	$C_{15}C_{11}O_2I_3$	63.05	63.34
4-Hydroxy-1,3-dimethylbenzene	125.0 - 126.4	47	$C_{15}H_{11}O_2I_3$	63.05	62.90
2-Hydroxy-1,4-dimethylbenzene	112.6–113.4°	55	$C_{15}H_{11}O_2I_3$	63.05	62.80
o-Hydroxydiphenyl	143.0 - 145.0	56	$C_{19}H_{11}O_2I_3$	58.39	58.50
<i>p</i> -Hydroxydiphenyl	169.8 - 170.6	64	$C_{19}H_{11}O_2I_3$	58.39	58.87
o-Iodophenol	$183.6 - 185.6^{\circ}$	70	$C_{13}H_6O_2I_4$	72.32	72.13
1-Naphthol	145.2 - 147.2	42	$C_{17}H_9O_2I_3$	60.82	60.82
2-Naphthol	$157.2 - 158.0^{\circ}$	33	$C_{17}H_9O_2I_3$	60.82	60.80
o-Nitrophenol	192.8 - 193.6	67	$C_{13}H_6O_4NI_3$	61.34	61.01
<i>m</i> -Nitrophenol	$176.2 - 177.2^{\circ}$	34	$C_{13}H_6O_4NI_3$	61.34	61.21
<i>p</i> -Nitrophenol	179.4 - 180.4	42	$C_{13}H_6O_4NI_3$	61.34	61.48
Phenol	$132.4 - 133.3^{\circ}$	71	$C_{13}H_7O_2I_3$	66.11	66.22
Pyrogallol-1,3-dimethyl ether	147.2 - 148.0	57	$C_{15}H_{10}O_{3}I_{3}$	59.87	59.74
Resorcinol monoethyl ether	106.0 - 106.8	29	$C_{15}H_{11}O_{3}I_{3}$	61.41	61.30
Thymol	98.6-100.6	29	$C_{17}H_{15}O_2I_3$	60.24	60.40
2,4,6-Trichlorophenol	$148.4 - 149.8^{\circ}$	46	$\mathbf{C}_{13}\mathbf{H}_4\mathbf{O}_2\mathbf{C}\mathbf{I}_3\mathbf{I}_2$	56.05	56.05

Table I. Phenol Esters of 2,3,5-Triiodobenzoic Acid

^a 50% Dioxane and 50% water. ^b Ethyl acetate. ^c Ethanol. ^d Benzene. ^c Toluene. ⁷75% Dioxane and 25% water.